Microscopía y Procesamiento de Imágenes

Microscopía para el Estudio de Biofilms Bacterianos

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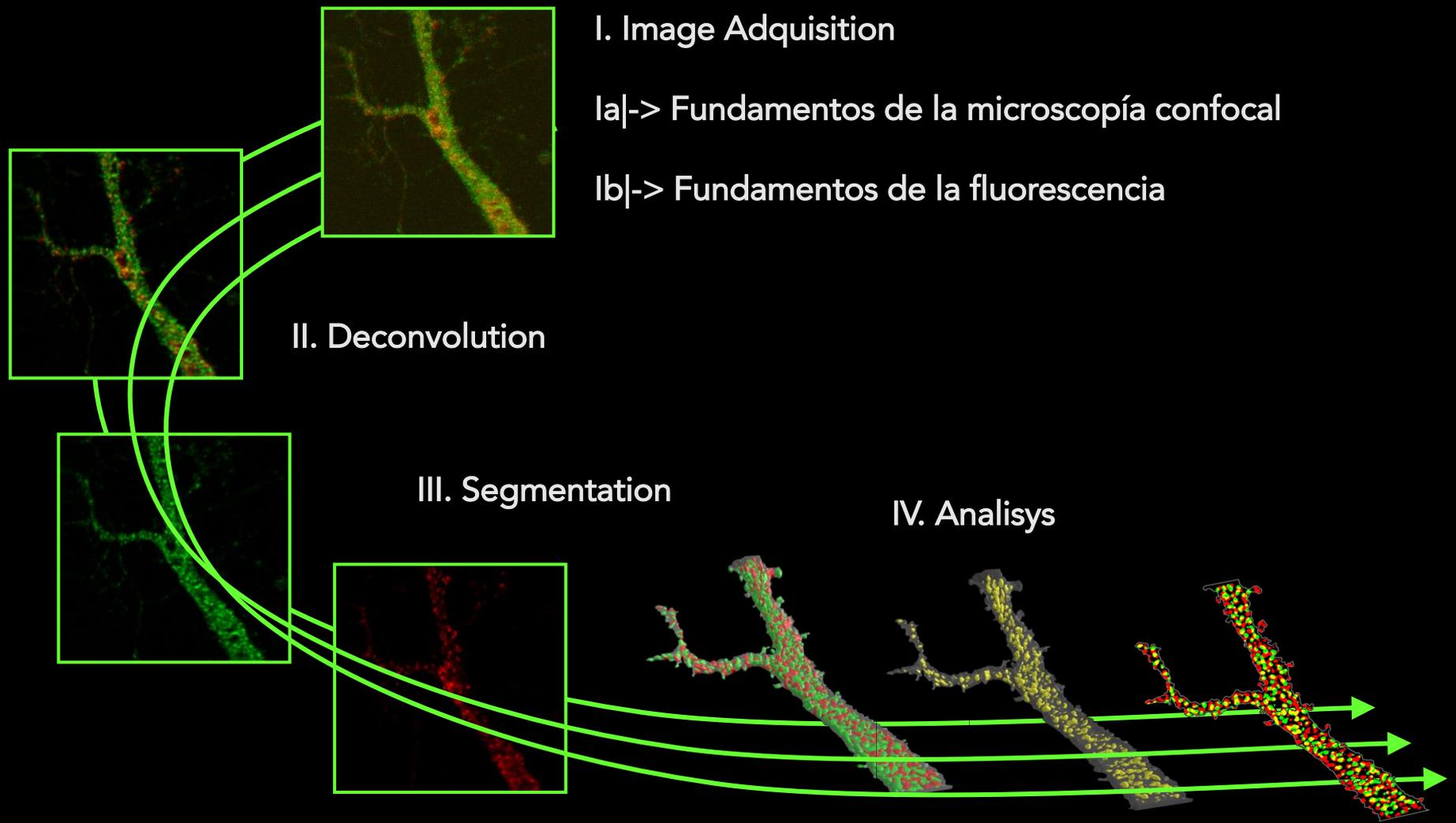Scientific Image Analysis

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The objective lens is the most critical component of a microscope and yet few researchers grasp the differences between specific objective classes.

Keep the acquisition settings constant between specimens to be compared quantitatively and particularly between sample and control.

"When you employ the microscope, shake off all prejudice, nor harbour any favorite opinions; for, if you do, 'tis not unlikely fancy will betray you into error, and make you see what you wish to see." Henry Baker, chapter 15, "Cautions in viewing objects" of The Microscope Made Easy, 1742.

"Remember that truth alone is the matter that you are in search after; and if you have been mistaken, let not vanity seduce you to persist in your mistake." Henry Baker, The Microscope Made Easy, 1742.

Welcome

I. Image Adquisition

Ia|-> Fundamentos de la microscopía confocal

Ib|-> Fundamentos de la fluorescencia

II. Deconvolution

III. Segmentation

IV. Analisys

Welcome

Richard Feynman (1918-1988)

Joaquín Torres García 1874-1949
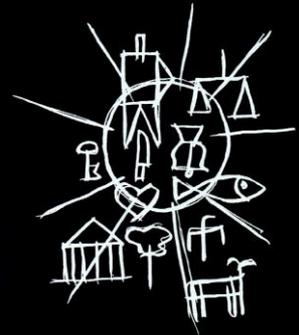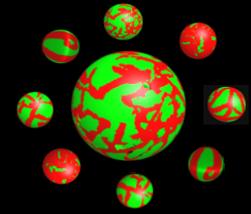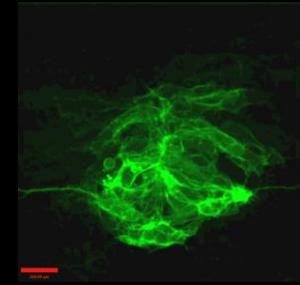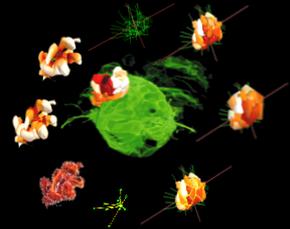
Maria Goeppert-Mayer 1906-1972

Mats Gustafson 2006 - 2011

René Descartes (1596-1650)

Ernst Abbe 1840- 2005

E Betzig   S Hell   W Moerner
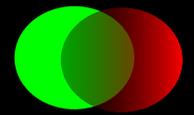
… just look at the thing …
¿ Human visual perception ?

Treatise of man (~ 1637)

Passions of the soul (~ 1649)

René Descartes (1596-1650)

glandula pinealis / **pineal organ**
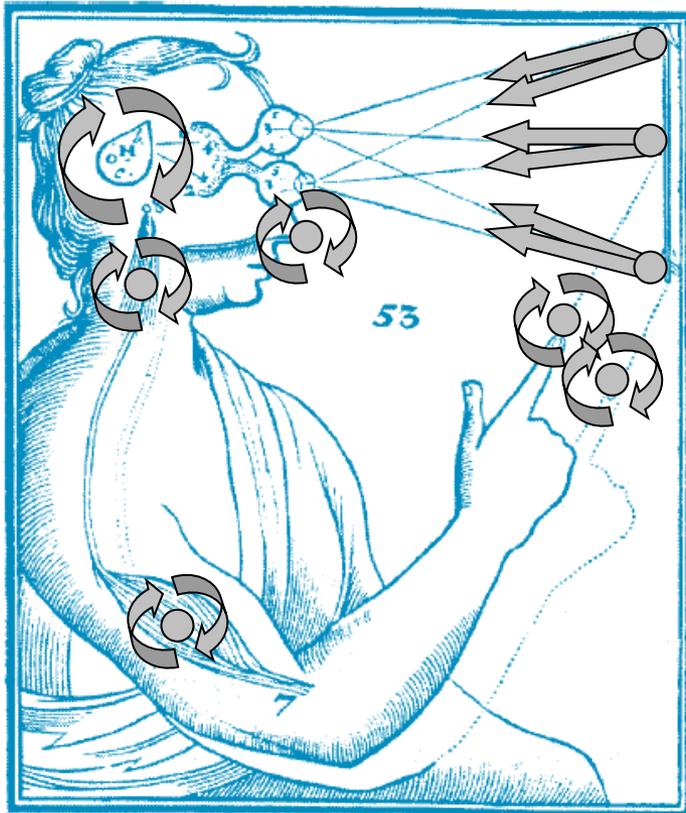
A combination of ...

1| direct signals ...

2| signals from other senses ...

3| feedback loops ...

... produce a symbolic representation of an object.

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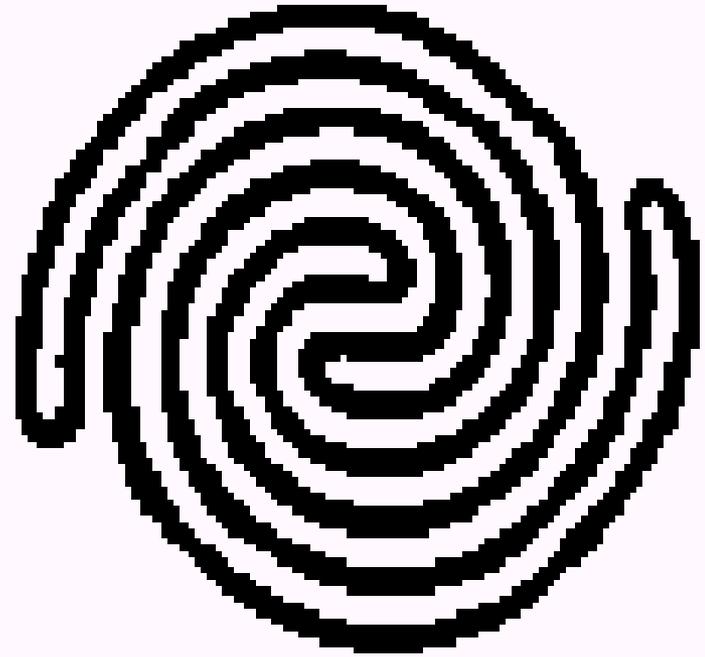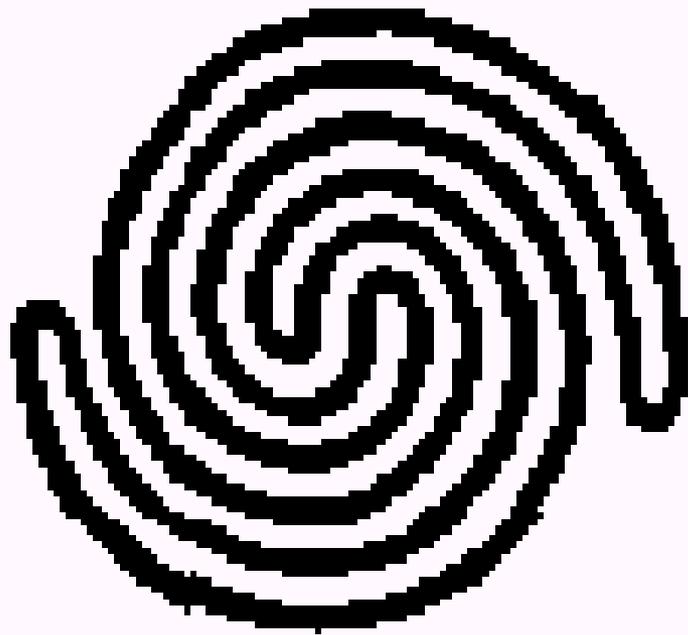
"It is very easy to answer many of these fundamental biological questions. You just look at the thing !

Make microscopes a hundred times more powerful and many problems of biology would be made very much easier."

Richard Feynman (1918-1988)

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OTF: Object/Optical Transfer Function
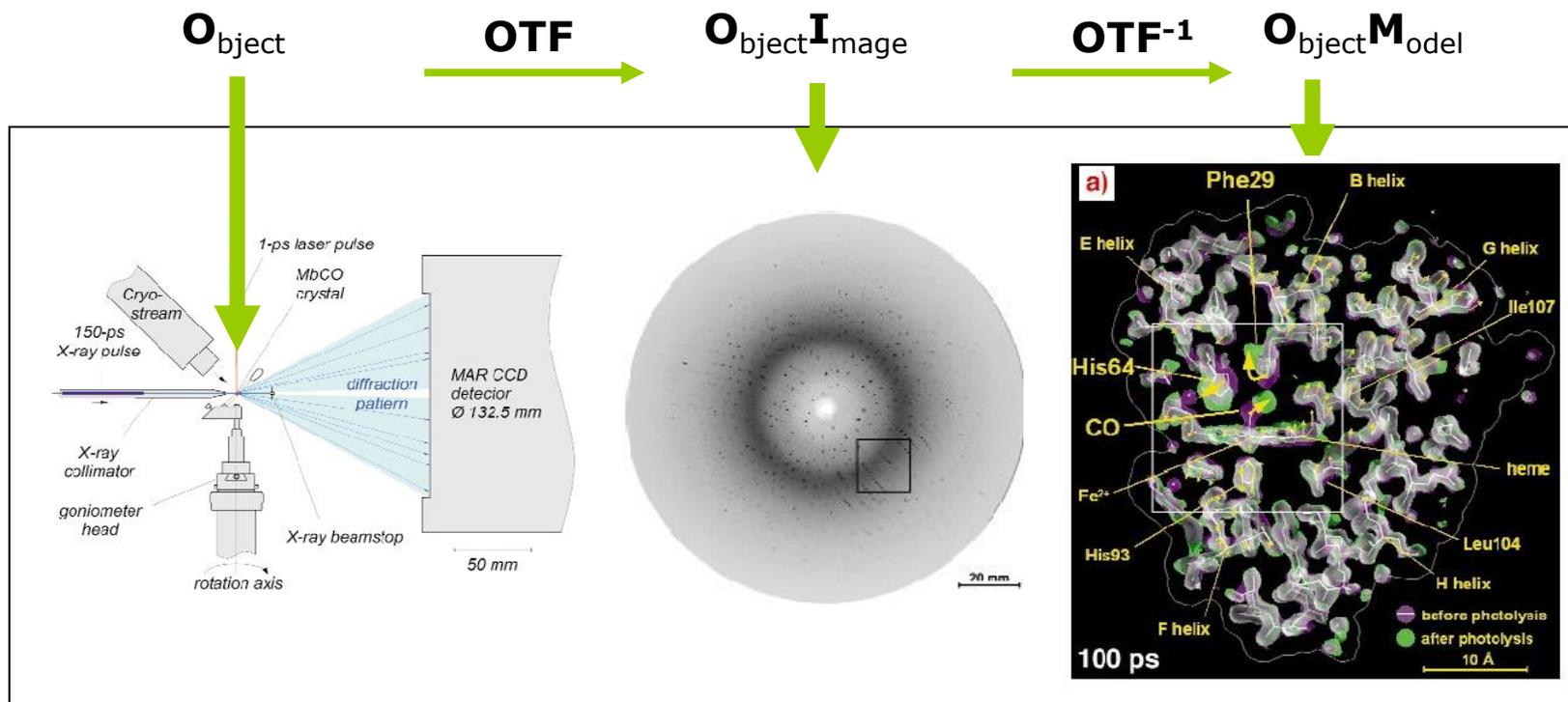Myoglobin in Action |Picosecond Laue Crystallography Diffraction Data
Schotte et al (2003) Science

O_{bject} **OTF** $O_{bject}I_{mage}$ **OTF^{-1}** $O_{bject}M_{odel}$

http://www.youtube.com/watch?v=lnKIBZYarzM

Diffraction Limited Resulution for a 10m telescope ~ λ/D ~ 0.01 arcsec
is limited to ~ 0.5 arcsec by the turbulent atmosphere.
NAOS creates an artificial star at 90 km altitude in the Earth´s mesosphere.
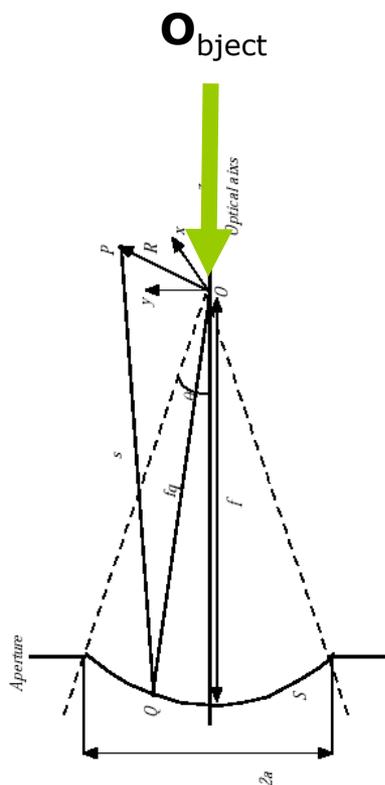The Laser Guide Star is used to correct atmospheric effects
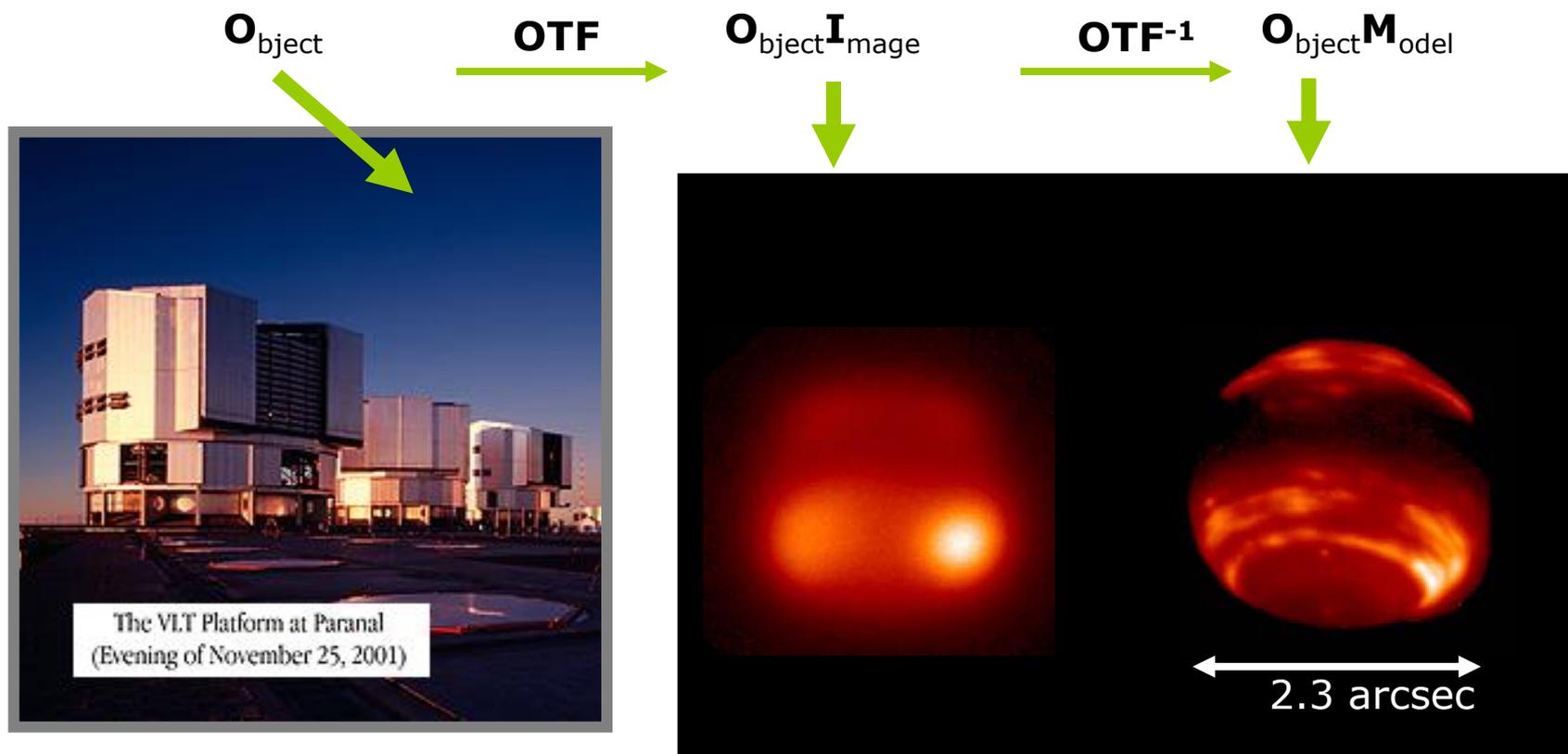
O_{bject} **OTF** → $O_{bject}I_{mage}$ **OTF^{-1}** → $O_{bject}M_{odel}$

The VLT Platform at Paranal
(Evening of November 25, 2001)

2.3 arcsec

Diffraction Limited Resulution for a 10m telescope ~ λ/D ~ 0.01 arcsec is limited to ~ 0.5 arcsec by the turbulent atmosphere.

$\mathbf{O}_{bject}$ **OTF** $\mathbf{O}_{bject}\mathbf{I}_{mage}$ **OTF^{-1}** $\mathbf{O}_{bject}\mathbf{M}_{odel}$

Hubble

VLT
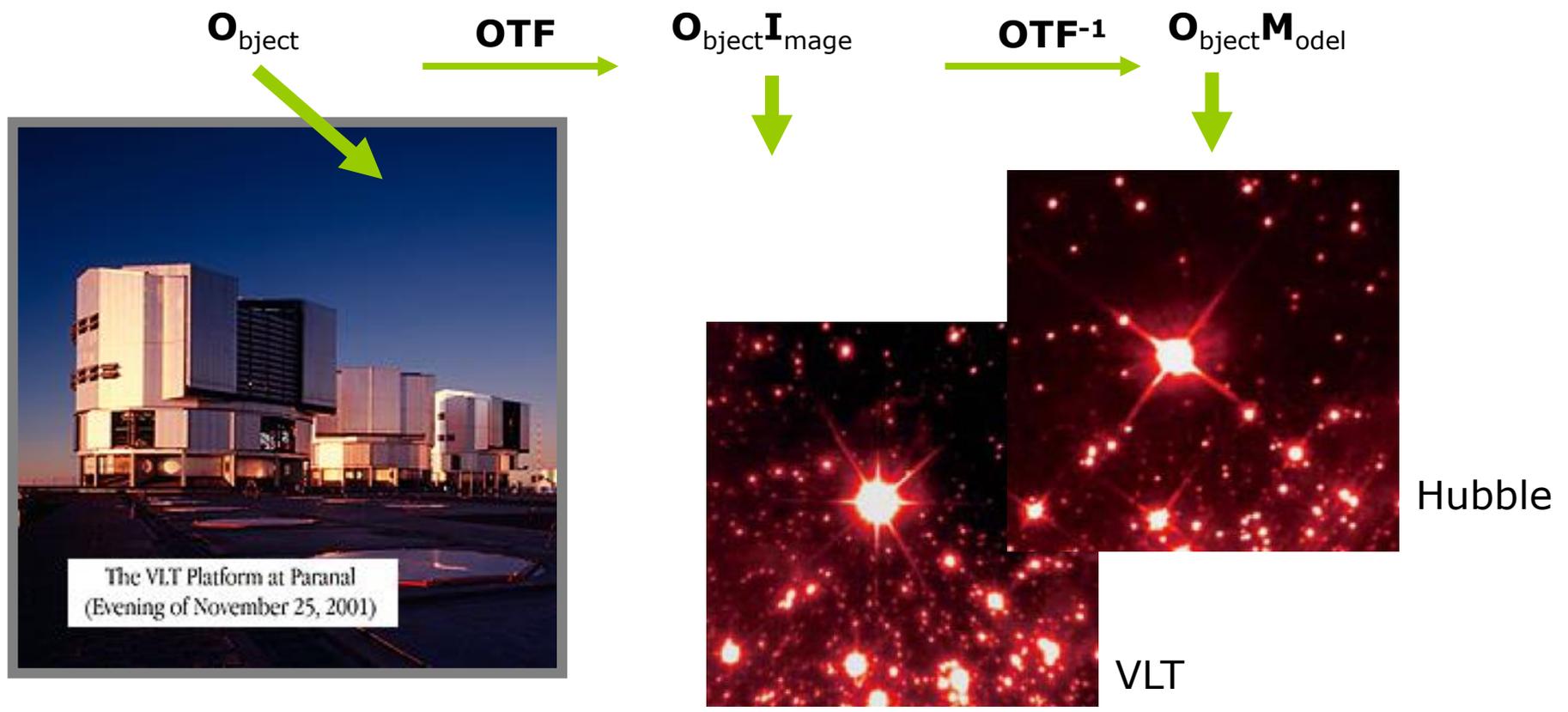
Confocal Microscopy | From Geometric Optics to Diffraction Theory

Diffraction: The deviation of an electromagnetic wavefront from the path predicted by geometric optics when the wavefront interacts with a physical object such as an opening or an edge.
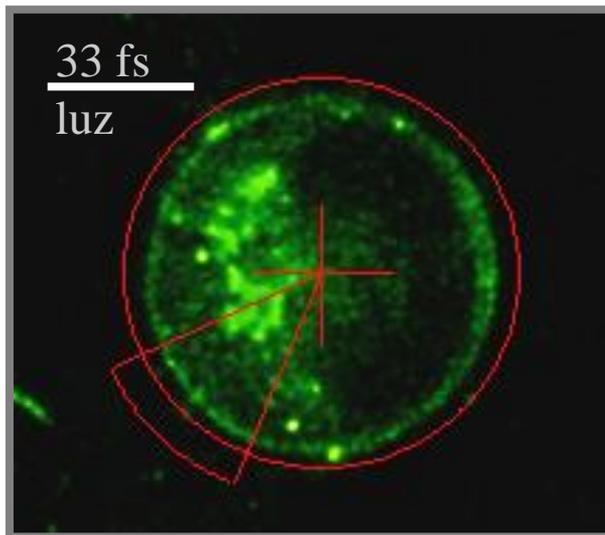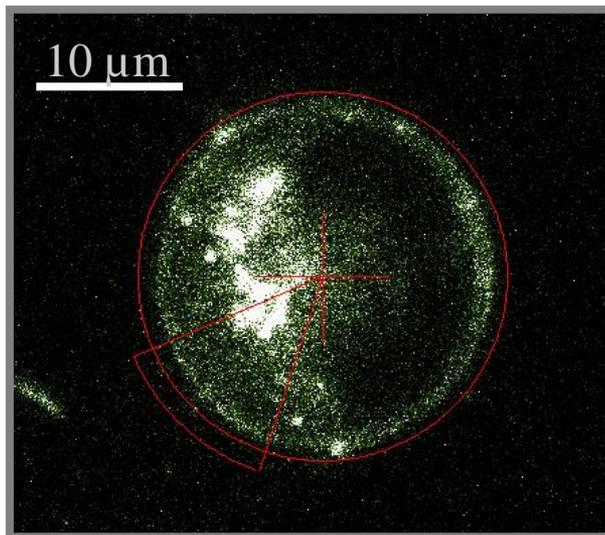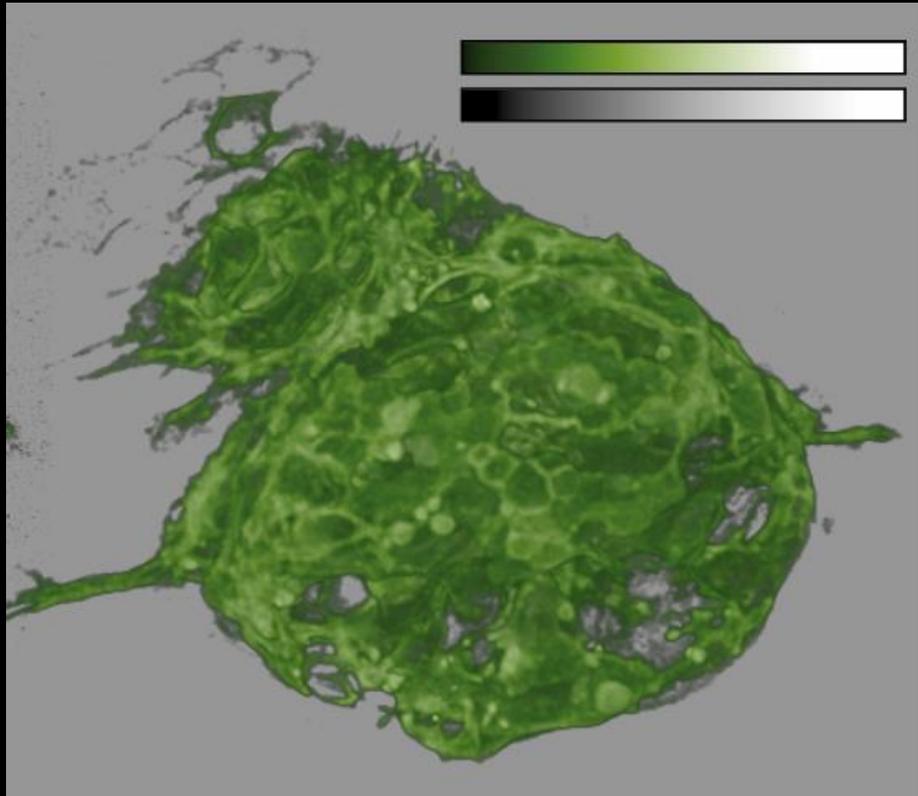
$\mathbf{O}_{bject}$ → **OTF** → $\mathbf{O}_{bject}\mathbf{I}_{mage}$ → **OTF^{-1}** → $\mathbf{O}_{bject}\mathbf{M}_{odel}$

10 μm

33 fs
luz
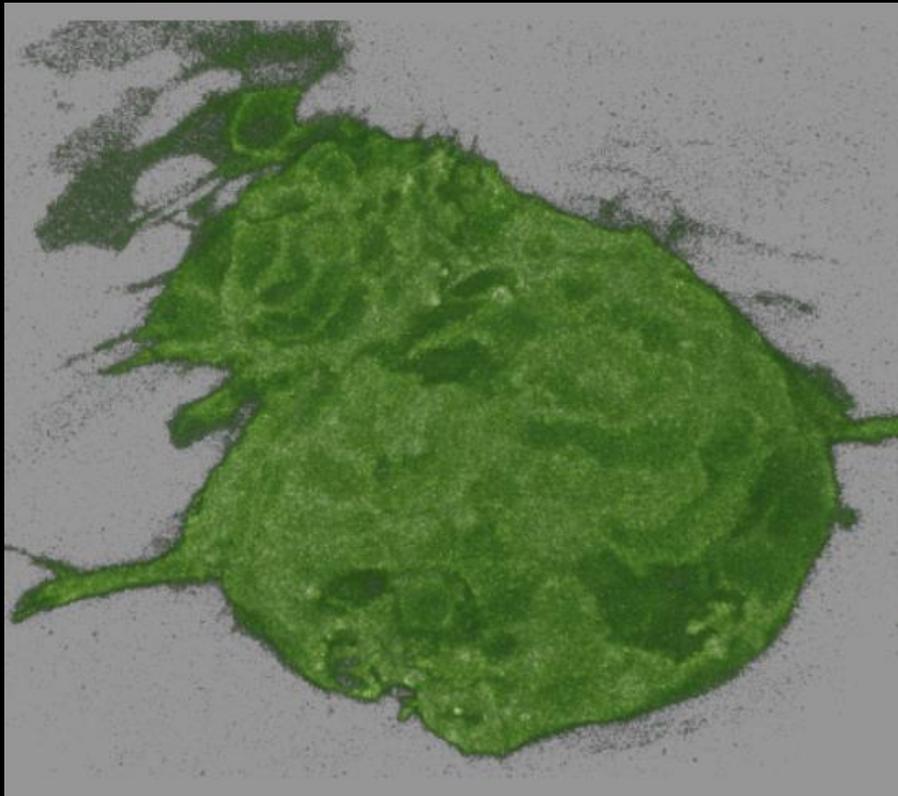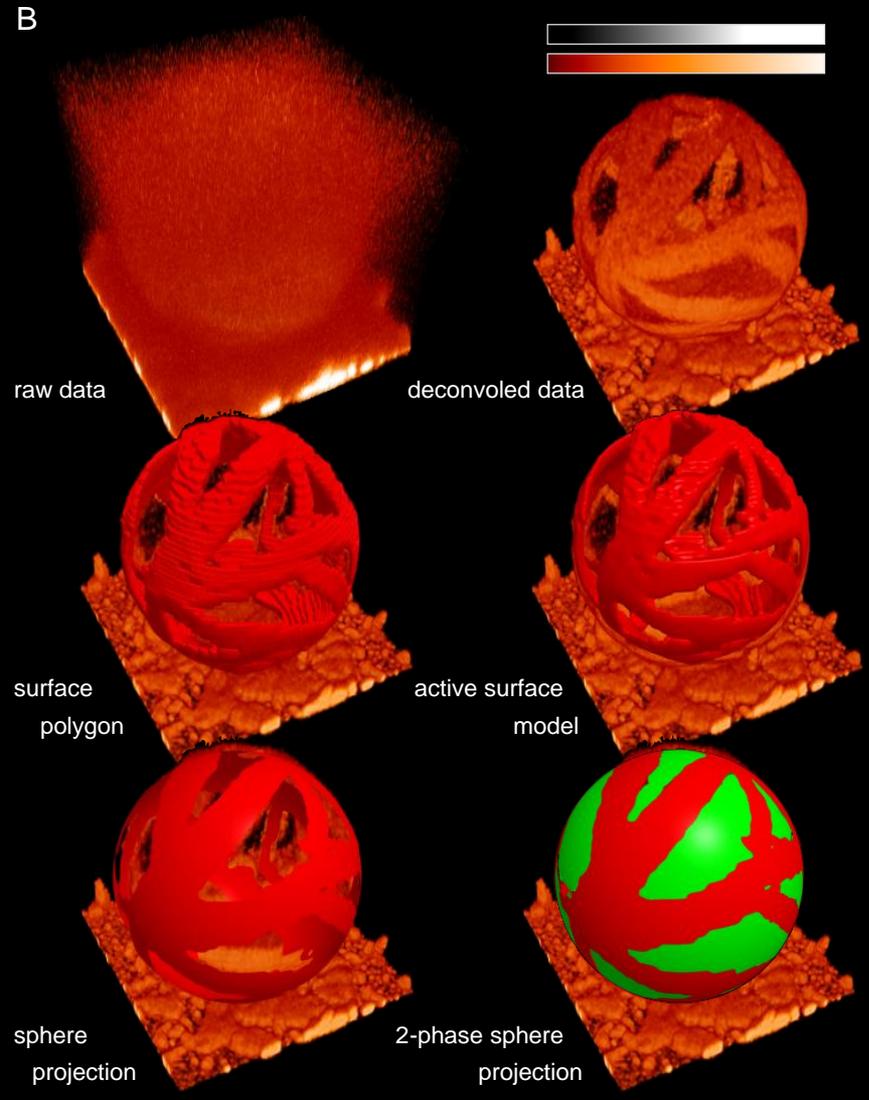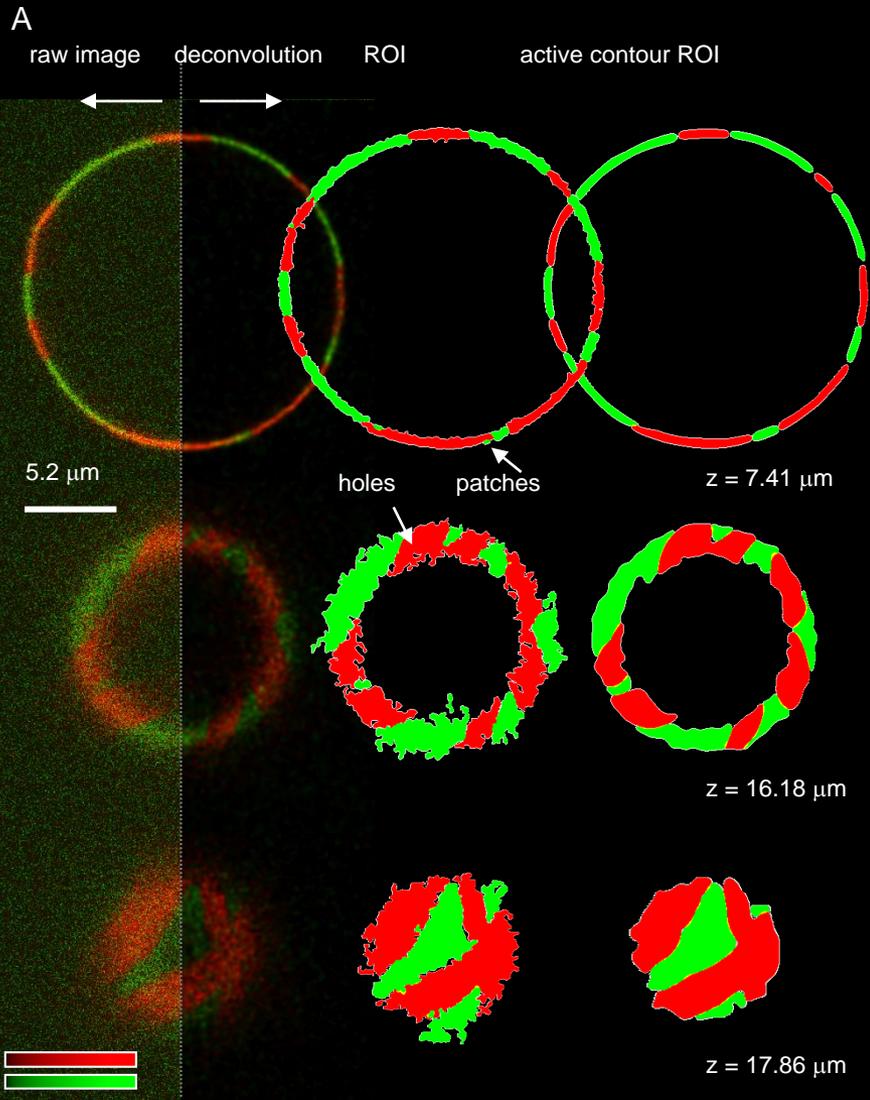
|-> Deconvolution

G. Schlapp et al. /
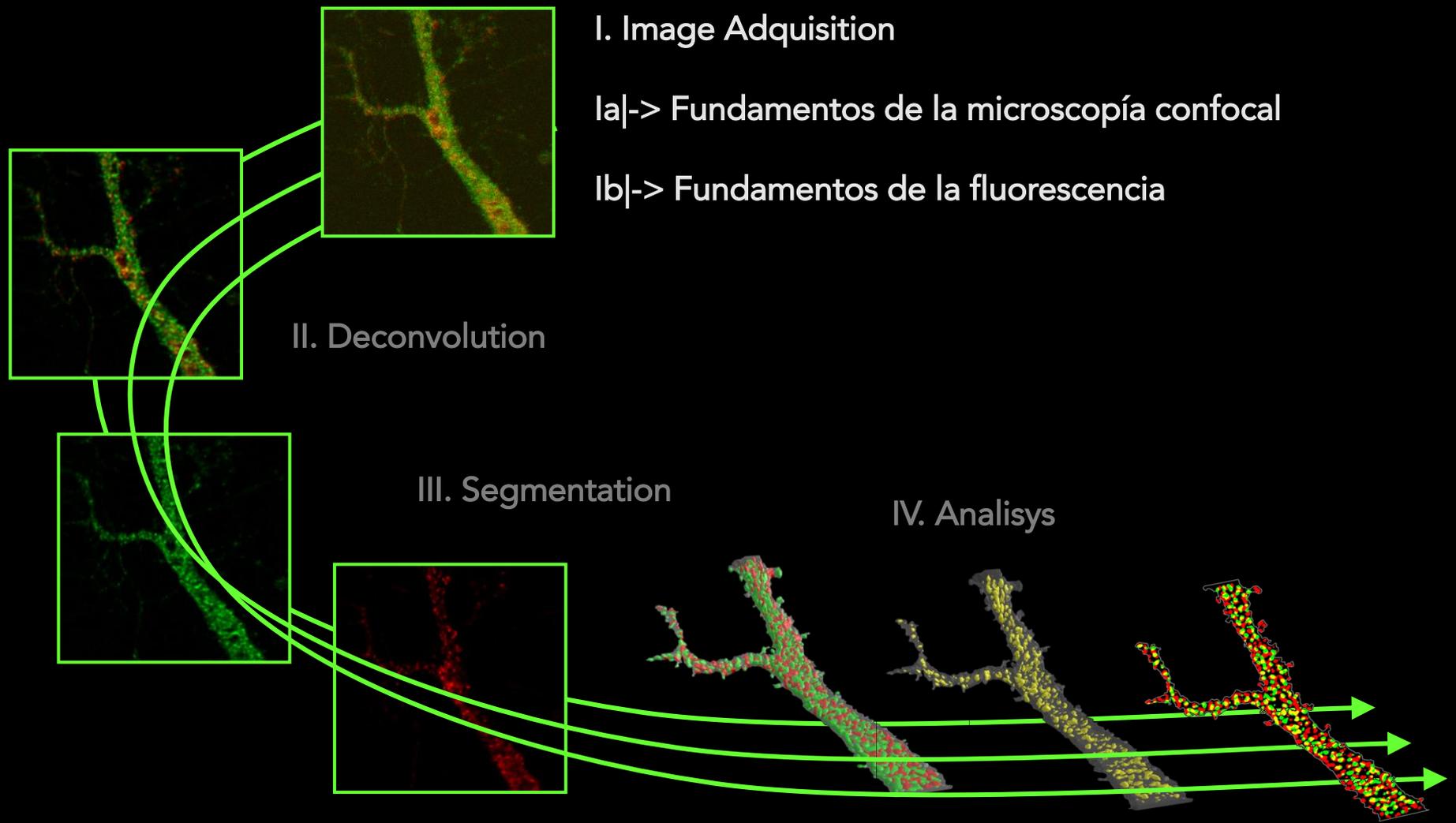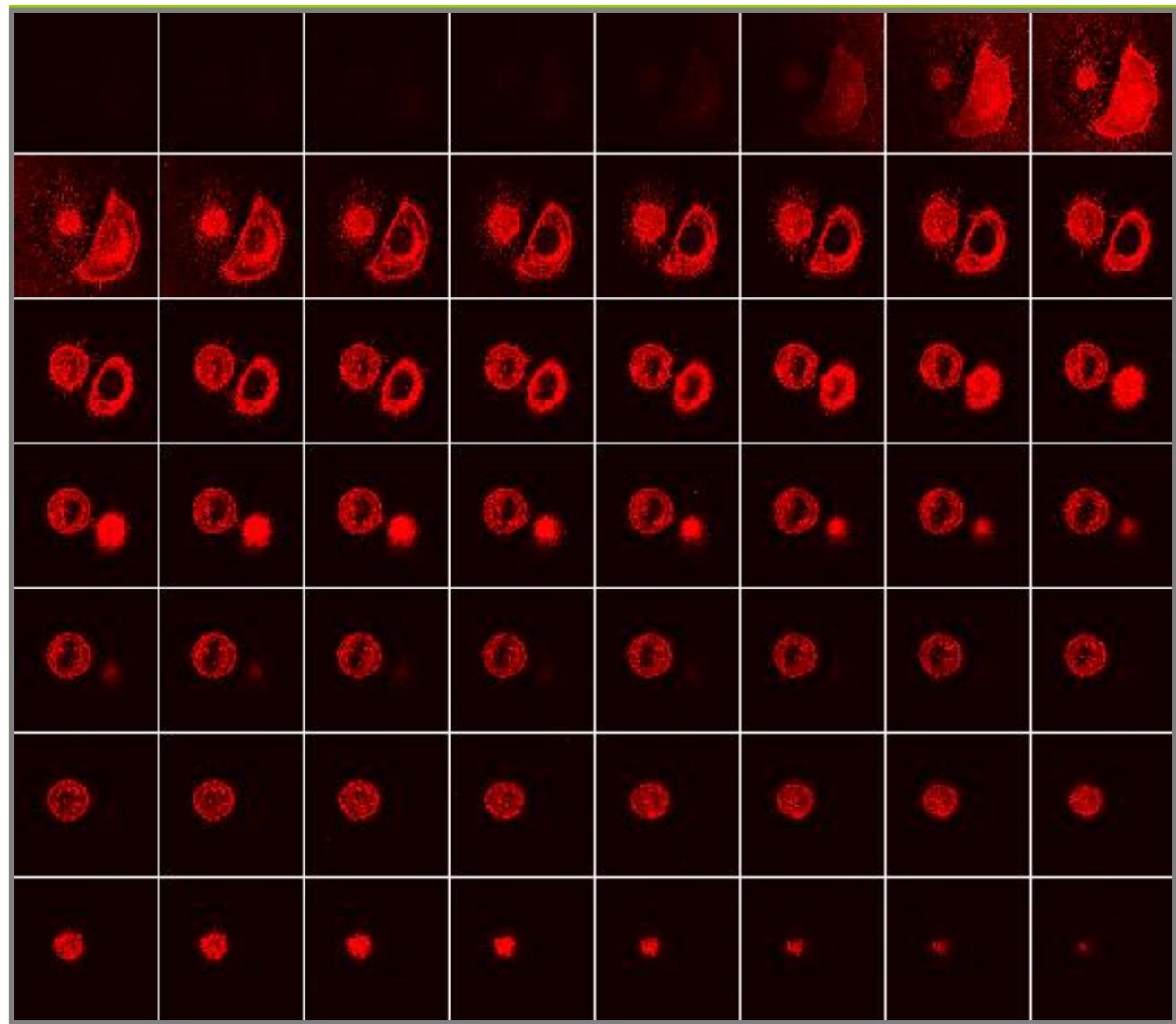Journal of
Microbiological
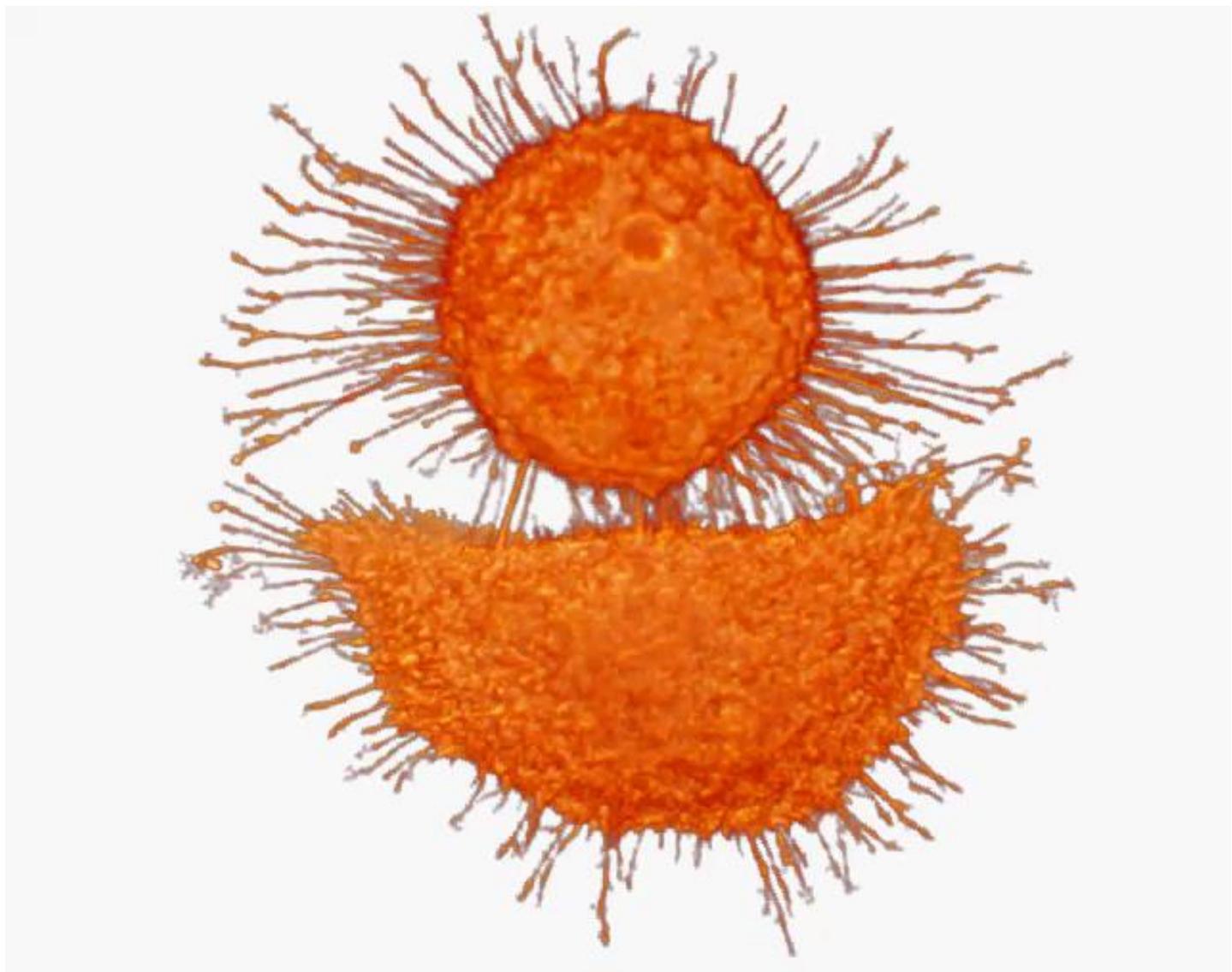Methods 87 (2011)
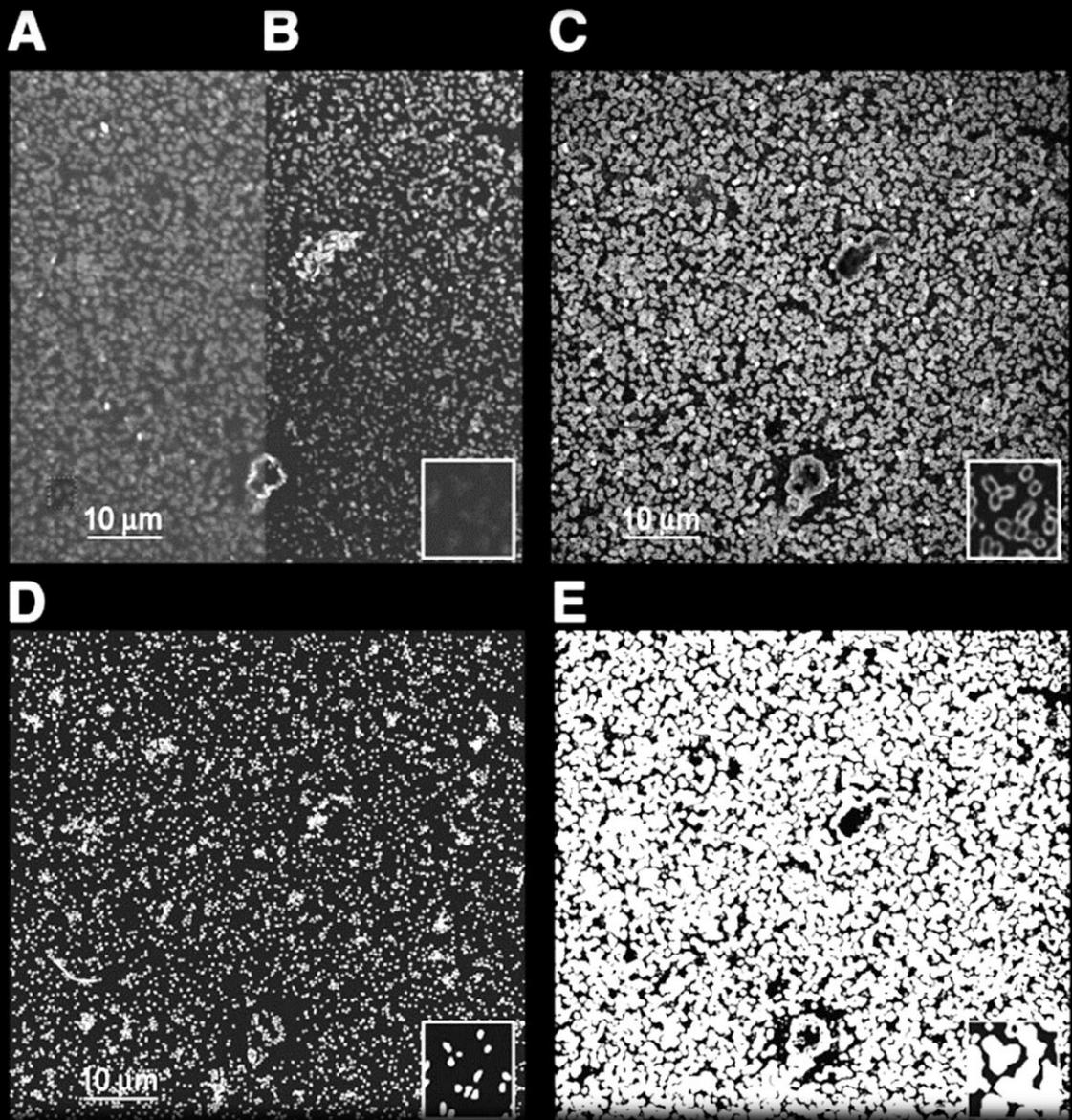
Hans Janssen (1595), ...
Galileo Galilei (1610)

|-> Microscopy

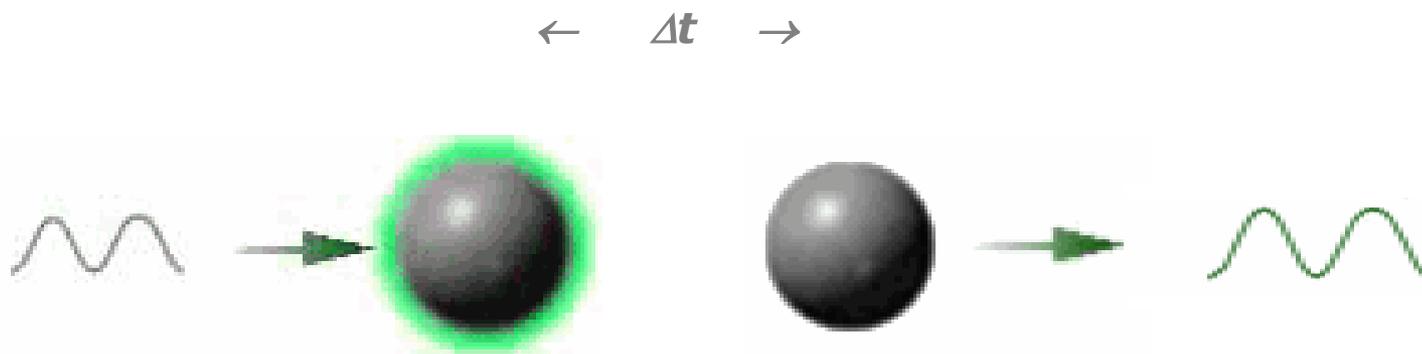**Luminescencia:**

- *Fluorescencia* $\Delta t \sim 10^{-8}s$
- *Fosforescencia* $\Delta t \sim 10^{-3}\text{-}10^{0}s$

Interacciones ...
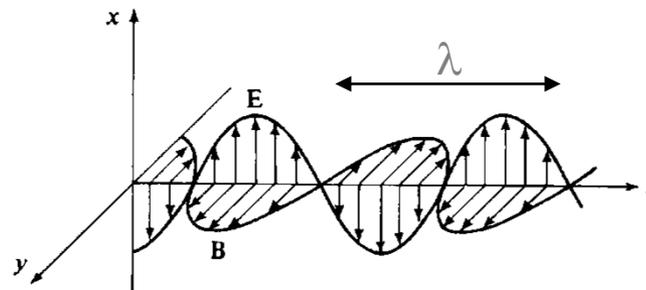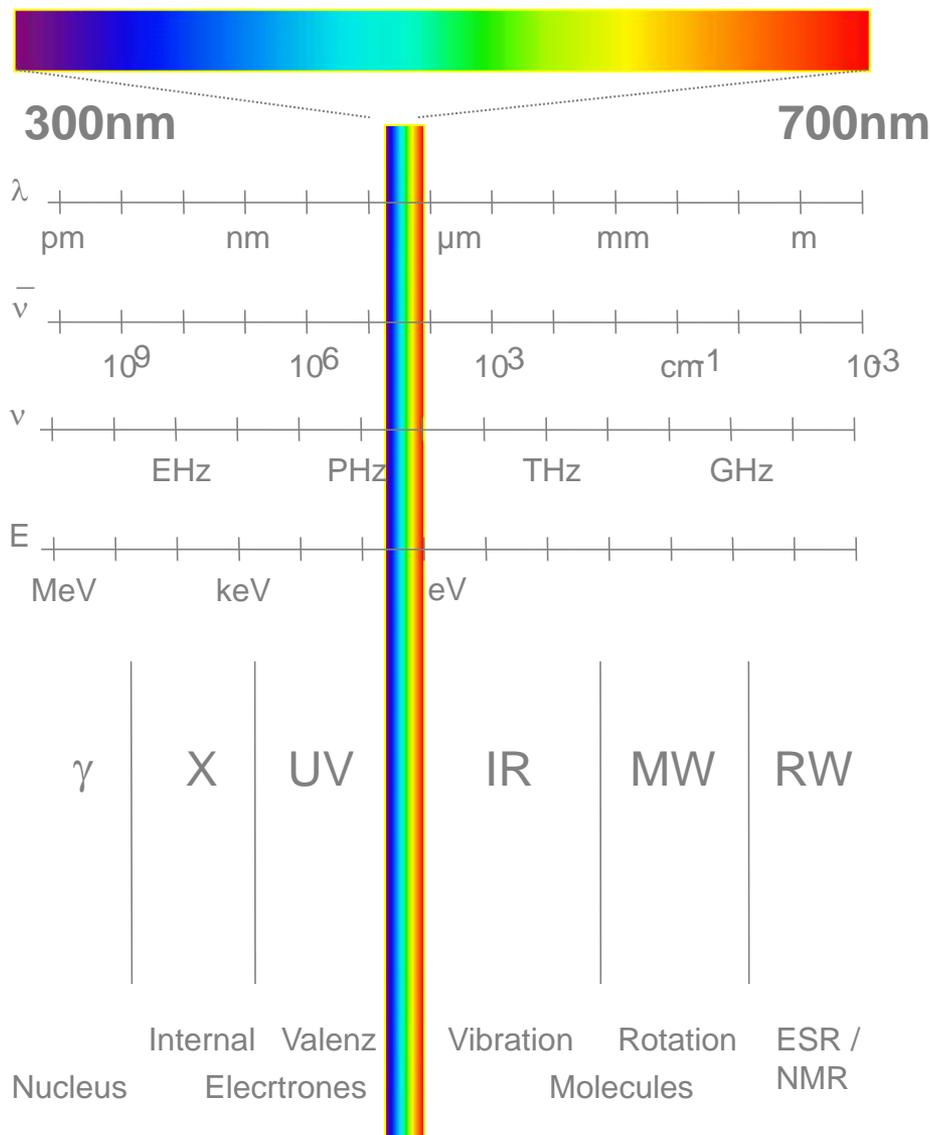- **intra- e inter moleculares ...**

producen cambios ...
- **espectrales**
- **tiempos de vida**
- **polarización**
- **intensidad ...**

$\leftarrow \quad \Delta t \quad \rightarrow$

Absorción / Excitación   *Emissión*

300nm   **700nm**

λ

pm   nm μm   mm   m

$\bar{\nu}$

10^9 10^6 10^3 cm^1 10^3

ν

EHz   PHz   THz   GHz

E

MeV   keV   eV

γ X   UV   IR   MW   RW

        Internal   Valenz   Vibration   Rotation   ESR / NMR

Nucleus   Elecrtrones   Molecules

Energía de un fotón: (~1-5eV)

$$E = h\,\nu = hc\lambda^{-1} \mid c = \nu\,\lambda$$

ν, frecuencia [s^{-1}]
h, constante de Planck [6.626 10^{-34} Js^{-1}]
λ, longitud de onda [m]
c, velocidad de luz [~3 10^8 ms^{-1}]

Energía molecular:

$$E = E_{rot} + E_{vib} + E_{el}$$
$$1 \quad : \quad 10^3 \quad : \quad 50 \cdot 10^3$$

Energía térmica:

$$E = k\,T \ (\sim 2.5\ 10^{-2}\ eV, T \sim 20°C)$$

k = Constante de Bolzmann (0.86 10^{-4} eV/K)
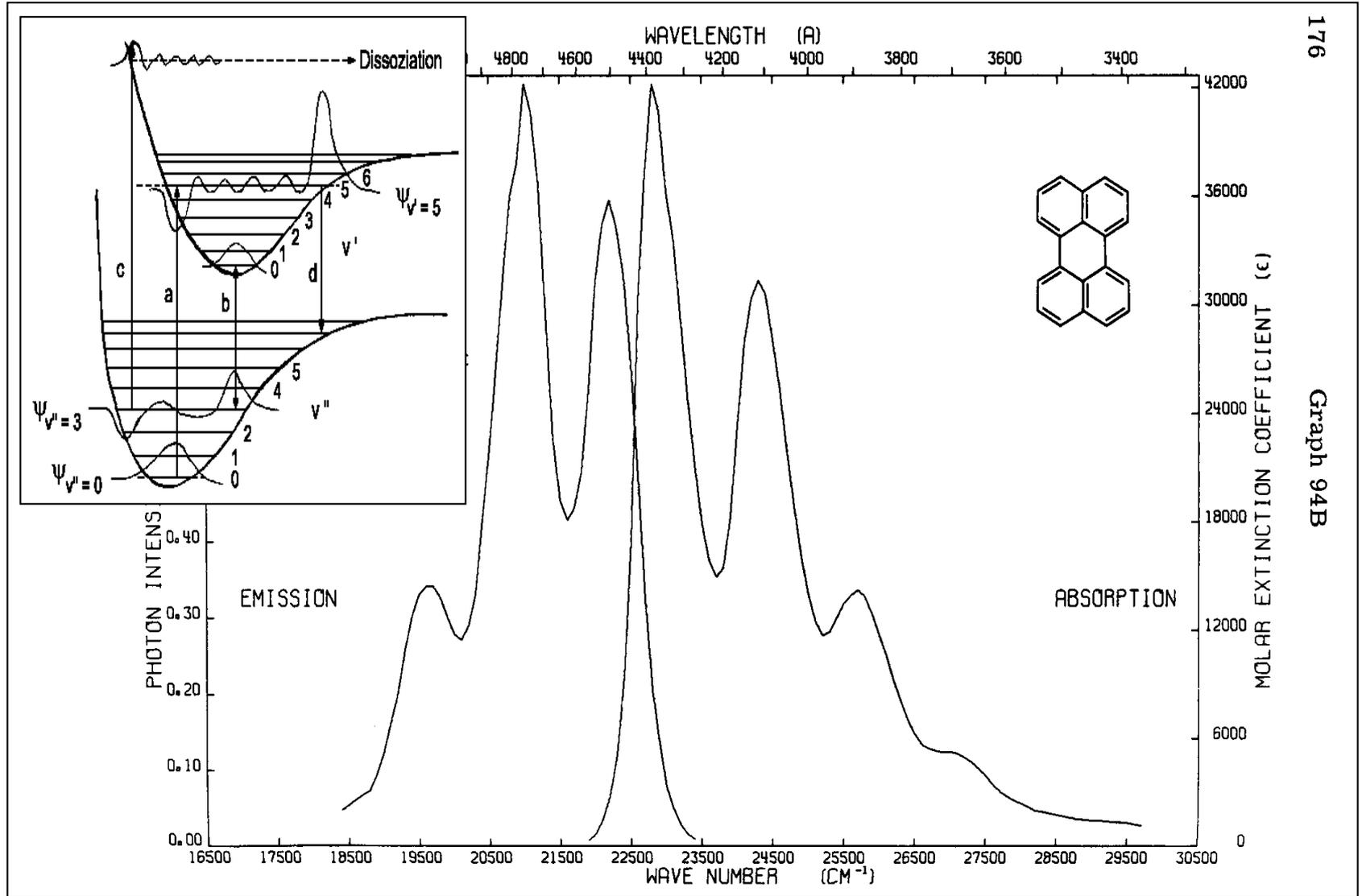
|-> Jablonski Diagram
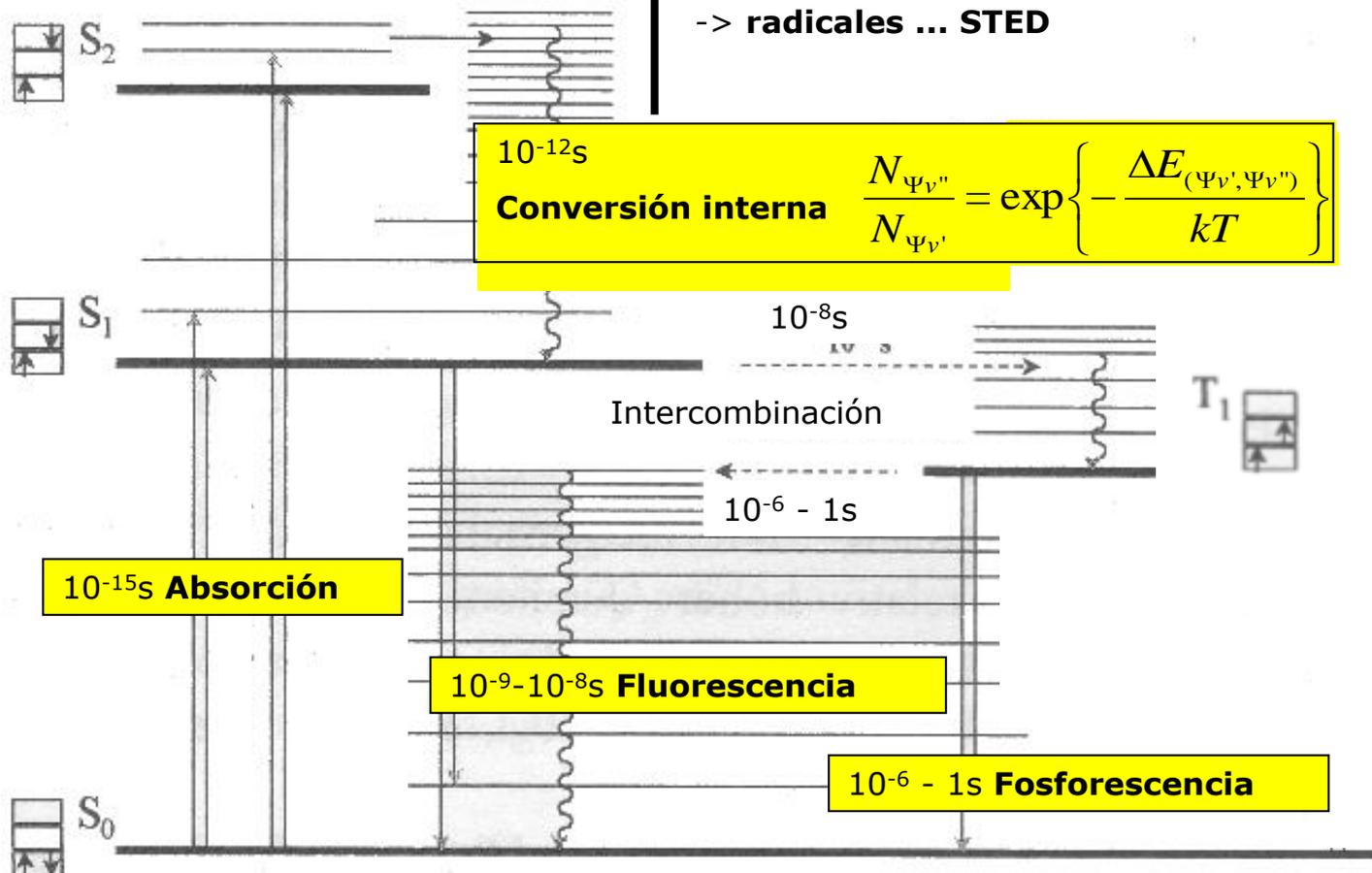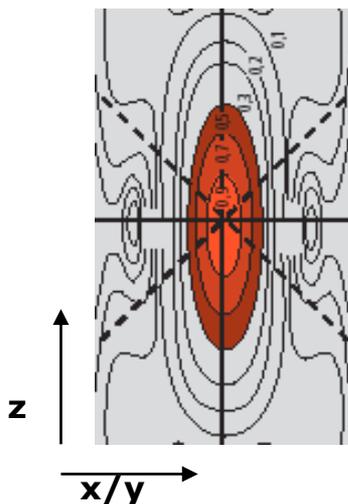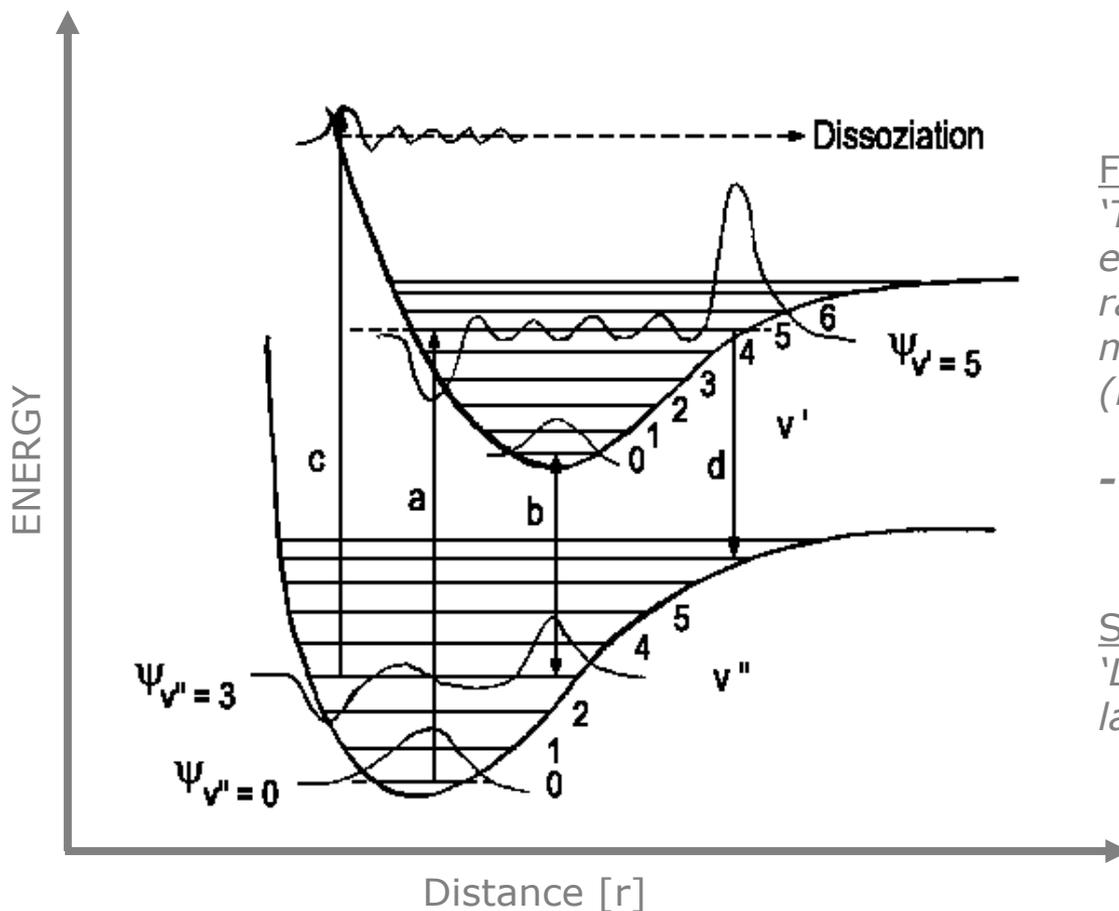
Frank Condon **:**
'Transiciones entre niveles electrónicos ocurren mucho mas rápido que movimientos de núcleos moleculares.'
$(masa_{electron} / masa_{atom}: 1 : 2000)$

-> Mirror Image Rule

Stokes (Shift) **:**
'La energía de emisión es menor a la energía de excitación.'

|-> Mirror Image Rule

- <u>Principles of Fluorescence Spectroscopy, Joseph R. Lakowicz 4.1 Introduction to Fluorescence</u>

|-> Quenching, Bleaching ...

|-> Polarisation ...

|-> Steady-State and Time-Resolved Fluorescence...

|-> Förster Resonance Energy Transfer ...